

# **Exploring Quantum Physics**

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# **Guest Lecture:** Electron Spin Part I: Zeeman effect



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## Photon emission spectrum of Hydrogen



### Degeneracy of spherically symmetric eigenstates

•Imagine we have a classical electron orbit with total angular momentum

$$\left|\hat{L}\right| = \sqrt{l(l+1)}\hbar$$

•This circulating charged particle comprises a current and so the orbit has a magnetic moment

$$\vec{\mu}| = I \cdot Area = -ef \cdot \pi r^{2} = -\frac{ev}{2\pi r} \cdot \pi r^{2}$$
$$= -\frac{evr}{2} = -\frac{e\hbar}{2m} \frac{mvr}{\hbar} = -\mu_{B} \frac{|\vec{L}|}{\hbar}$$
$$\mu_{B}: \text{"Bohr magneton"}$$

• Therefore, the electron has additional energy in magnetic field:

$$E = -\vec{\mu} \cdot \vec{B}$$
  $E = -\mu_z B_z$  (for  $\vec{B} = B_z \hat{Z}$ )

• So, if different states with the same principle quantum number n have different < $\mu_z$ >, their energy eigenvalues will shift differently in a magnetic field and the degeneracy will be broken!





• So the energy added to each state due to the magnetic field is

$$E = -\langle \mu_z \rangle B_z = \mu_B \frac{\langle L_z \rangle}{\hbar} B_z = \mu_B m B_z \qquad \text{(for } \vec{B} = B_z \vec{Z})$$

• Note that  $\mu_B$ =5.8 x 10<sup>-5</sup> eV/T (small in comparison to electronic transitions even for large magnetic fields)

### "Normal" Zeeman effect



- In a magnetic field, the otherwise degenerate  $l \neq 0$  levels split into "multiplets" separated by "Zeeman" energy
- What effect does this have on the observed optical spectra? i.e. what did Zeeman see?
- Not all transitions are allowed!
- We must derive "selection rules"



Pieter Zeeman Nobel Prize, 1902

## **Electronic Dipole Transitions**

$$\psi_{nlm} \rightarrow \psi_{n'l'm'}$$
initial  $\rightarrow$  final

• During transition, electron is in a superposition state

$$\Psi \sim \psi_{nlm} e^{-irac{E_n}{\hbar}t} + \psi_{n'l'm'} e^{-irac{E_m}{\hbar}t}$$

• This charged electron "cloud" has a dipole moment

$$\begin{aligned} -e\langle \vec{r} \rangle &= -e\langle \Psi | \vec{r} | \Psi \rangle \\ &= -e \int \vec{r} \{ |\psi_{nlm}|^2 + |\psi_{nlm}|^2 + \psi^*_{n'l'm'} \psi_{nlm} e^{-i\omega t} + \psi^*_{nlm} \psi_{n'l'm'} e^{+i\omega t} \} d^3 r \\ &= -e \int \vec{r} \{ \psi^*_{n'l'm'} \psi_{nlm} e^{-i\omega t} + \psi^*_{nlm} \psi_{n'l'm'} e^{+i\omega t} \} d^3 r \end{aligned}$$



• So the only transitions allowed have  $\Delta m=0,+1$ , or -1. This is our "selection rule"

• Also, from the symmetry of the polar angle  $\theta$  integral, we have another selection rule  $\Delta \ell = +1$  or -1 (without proof)

#### "Normal" Zeeman selection rules

 $\Delta \ell$ =+1 or -1 and  $\Delta m$ =0,+1, or -1.



• Each spectral line splits into a triplet

• By the uncertainty principle  $\Delta t \Delta E^{-h}$ ,  $\Delta E$  (spectral linewidth) is inversely proportional to excited state lifetime  $\Delta t$ 

• Note that 2s-1s is forbidden by dipole selection rules. Since the 2s lifetime  $\Delta t$  is large, the spectral linewidth  $\Delta E$  is (very) small! This enables precision measurements at 1 part in 10<sup>15</sup> such as the "Lamb shift": experimental verification of quantum electrodynamical effects



Willis Lamb Nobel Prize, 1955



 $\Delta m=-1 \Delta m=0 \Delta m=+1$ 

Nobel Prize, 1902

#### "Anomalous" Zeeman effect

- More commonly we see multiplet splitting other than 3!
- What are we forgetting?
- Go back to assumptions:

$$-\frac{\hbar^2}{2m}\nabla^2\Psi = i\hbar\frac{\partial}{\partial t}\Psi \quad \text{(free particle)}$$

- This uses the nonrelativistic kinetic energy  $E=1/2mv^2=p^2/2m$
- Note this equation does not treat time and space on an equal footing: 1<sup>st</sup> derivative for time and 2<sup>nd</sup> derivative for space
- More correctly we should use the relativistic expression for kinetic energy

$$E = \sqrt{(mc^2)^2 + (pc)^2}$$

• Note that this is asymptotically equivalent to classical expression:

$$E = mc^2 \sqrt{1 + \left(\frac{pc}{mc^2}\right)^2} \approx mc^2 \left(1 + \frac{1}{2} \left(\frac{pc}{mc^2}\right)^2 + \cdots\right) = mc^2 + \frac{p^2}{2m} + \text{small relativistic corrections}$$



#### **Relativistic Quantum Mechanics**

• Interpret total energy expression as an operator to construct wave equation

$$\sqrt{(mc^2)^2 + (pc)^2}\Psi = i\hbar \frac{\partial}{\partial t}\Psi$$
 (free particle)

• But how to interpret the action of an operator from within a square root??

• This problem disappears if the expression inside is a perfect square... is it??

$$(mc^{2})^{2} + (pc)^{2} = (\alpha_{0}mc^{2} + \alpha_{1}p_{1}c + \alpha_{2}p_{2}c + \alpha_{3}p_{3}c)^{2}$$
$$= (\alpha_{0}mc^{2} + \sum_{j=1}^{3}\alpha_{j}p_{j}c)^{2}$$

Only if  $\alpha_i^2 = 1$  and  $\alpha_i \alpha_i + \alpha_i \alpha_i = \{\alpha_i, \alpha_i\} = 0$  for  $i \neq j$ : This defines a "Clifford algebra"



Paul Dirac Nobel Prize, 1932

$$\left(\alpha_0 m c^2 + \sum_{j=1}^3 \alpha_j p_j c\right) \Psi = i\hbar \frac{\partial}{\partial t} \Psi \qquad \text{``Dirac Equation''}$$

• Note that this puts space and time on equal footing as required by relativity (only first derivatives)

## "Irreducible representation" of Clifford Algebra

- Must use matrices but how big?
- smallest which satisfy algebra are 4x4:

$$\alpha_0 = \begin{bmatrix} I_2 & 0_2 \\ 0_2 & -I_2 \end{bmatrix} \qquad \alpha_j = \begin{bmatrix} 0_2 & \sigma_j \\ \sigma_j & 0_2 \end{bmatrix} \qquad j = 1, 2, 3$$

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•Pauli matrices

$$\sigma_1 = \sigma_x = \begin{bmatrix} 0 & 1 \\ 1 & 0 \end{bmatrix} \qquad \qquad \sigma_2 = \sigma_y = \begin{bmatrix} 0 & -i \\ i & 0 \end{bmatrix} \qquad \qquad \sigma_3 = \sigma_z = \begin{bmatrix} 1 & 0 \\ 0 & -1 \end{bmatrix}$$

$$\begin{bmatrix} mc^2 I_2 & \vec{\sigma} \cdot \vec{p}c \\ \vec{\sigma} \cdot \vec{p}c & -mc^2 I_2 \end{bmatrix} \Psi = i\hbar \frac{\partial}{\partial t} \Psi \qquad (\vec{\sigma} = \sigma_x \hat{x} + \sigma_y \hat{y} + \sigma_z \hat{z})$$

• Note that when p=0, eigenvalues are +mc<sup>2</sup> and -mc<sup>2</sup> (twofold degenerate)

• Degeneracies are a signature of symmetry... but which one? What degree of freedom do these two values correspond to? Look to experiment for a clue...